

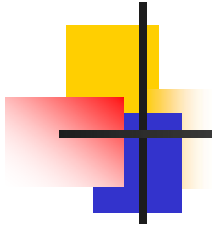
# Overview of NH<sub>3</sub> Emission from Poultry Facilities and the BMPs and BATs

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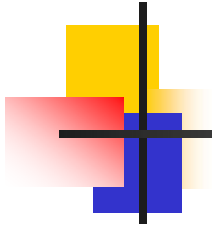




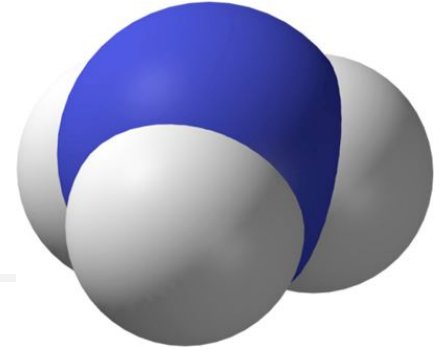
# Outline

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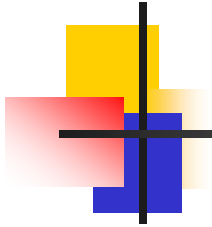
- Ammonia ( $\text{NH}_3$ ) and its generation at animal farms
- $\text{NH}_3$  emissions from poultry facilities
- Its health and environmental impacts
- Overview of the BMPs and BATs
- Summary



# What is Ammonia (NH<sub>3</sub>)?



- A gas (NH<sub>3</sub>), colorless, light than air, highly soluble in water, and has a sharp pungent odor detectable at 5 - 18 ppm.
- Ammonia has two forms, gas phase and liquid or solid ammonium phase.
- one of the important forms of nitrogen (NH<sub>3</sub>, NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, N<sub>2</sub>O, NO<sub>x</sub> and atmospheric N<sub>2</sub>) participating in the nitrogen cycle.
- Reacts with gaseous nitrate and sulfate for form fine particles.
- used for the production of fertilizers, nitric acids, fuels, explosives, and refrigerants;

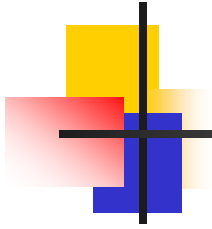


# Ammonia Gas and Ammonium

Ammonia easily convert from gaseous phase to liquid and solid phase ammonium ( $\text{NH}_4^+$ ) as pH changes.



- pH=9.25 (>7)
  - ★ 50% as un-ionized ( $\text{NH}_3$ )
  - ★ 50% as ionized ( $\text{NH}_4^+$ )
- pH<7
  - ★ 99% as ionized ( $\text{NH}_4^+$ )



# How $\text{NH}_3$ is Generated?

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- ◆  $\text{NH}_3$  is generated because inefficient conversion of feed N to animal products resulted in N excretion in urine of pigs and cattle and in the uric acid of poultry.
- ◆ The fecal enzyme urease catalyzes the hydrolysis of urine urea to form ammonia gas.
- ◆ Aerobic decomposition of uric acid of poultry also form ammonia gas.
- ◆ Mineralization of undigested protein also result in ammonia gas emission



## How $\text{NH}_3$ is Generated (cont)?

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- ★ Urine (swine and cattle)

- Urea hydrolysis



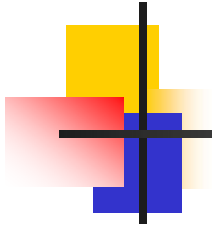
- ★ Uric acid (poultry)

- Aerobic decomposition of uric acid



- ★ Undigested protein

- Mineralization  $\rightarrow \text{NH}_3$

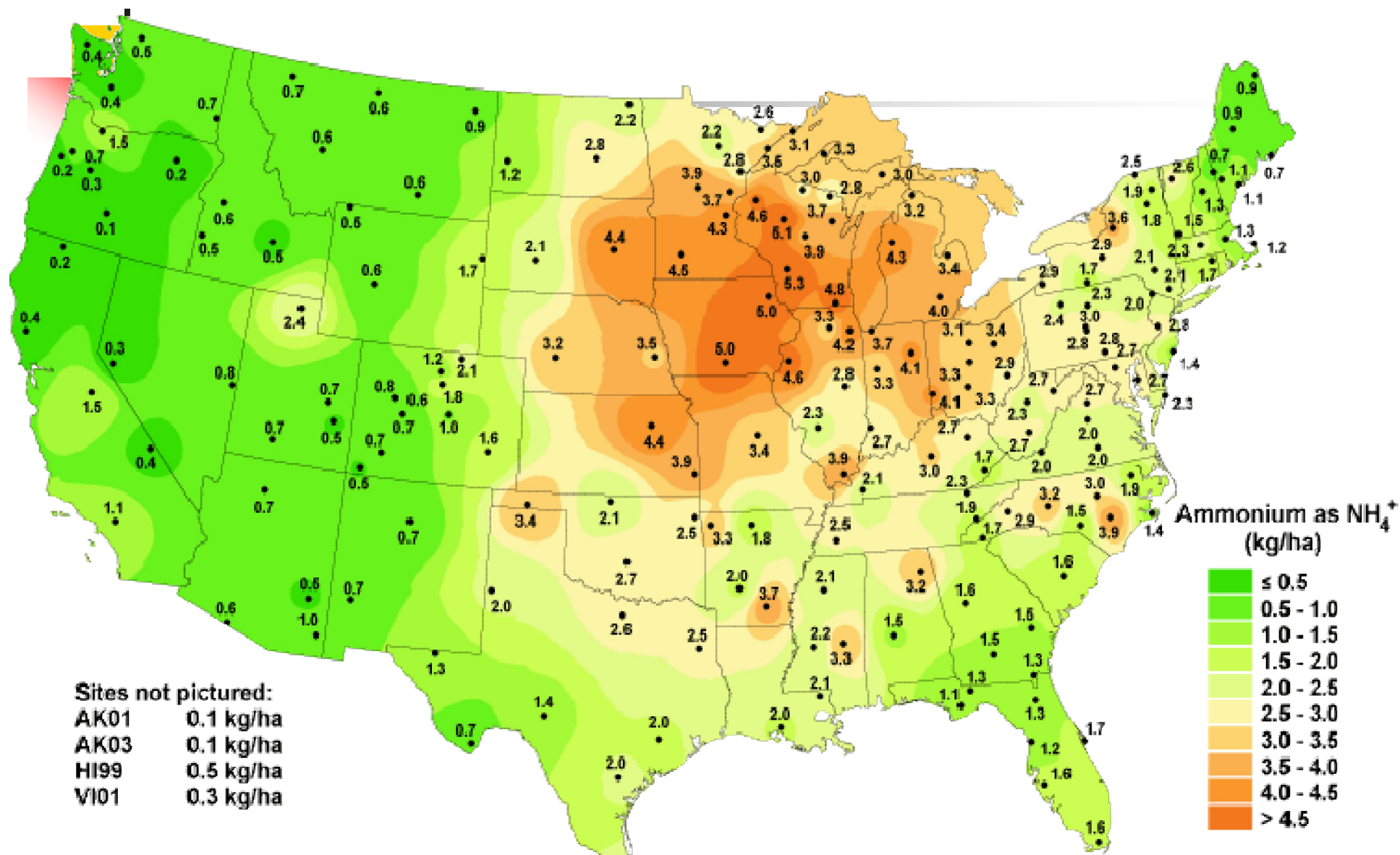


## Fate and Transport of $\text{NH}_3$ Cont...

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- ★ After released from the sources, ammonia will disperse vertically and horizontally in the atmosphere.
- ★ Dry and wet deposition will happen near the sources.
- ★ It has a lifetime of 1-5 days in the atmosphere.
- ★ Chemical reactions with other chemicals in the atmosphere form fine particles.
- ★ When is converted to  $\text{NH}_4^+$  aerosols, the lifetime increases up to 15 days.

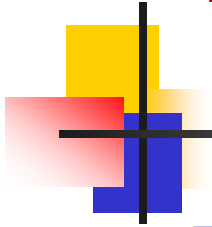
# Ammonium ion wet deposition, 2001



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.sws.uiuc.edu>



# Factors Affecting $\text{NH}_3$ Emission and Transport



## ■ Dietary Factors

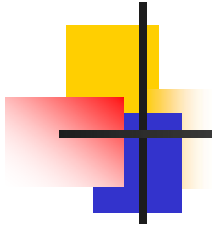
- ★ Excess crude protein in diet
- ★ Low conversion of dietary N to animal products
- ★ Total nitrogen content

## ■ Environmental Factors

- ★ pH
- ★ Temperature
- ★ Moisture content
- ★ Chemical and microbiological activities--additives
- ★ Ammonia concentration of air
- ★ Ventilation air / wind speed

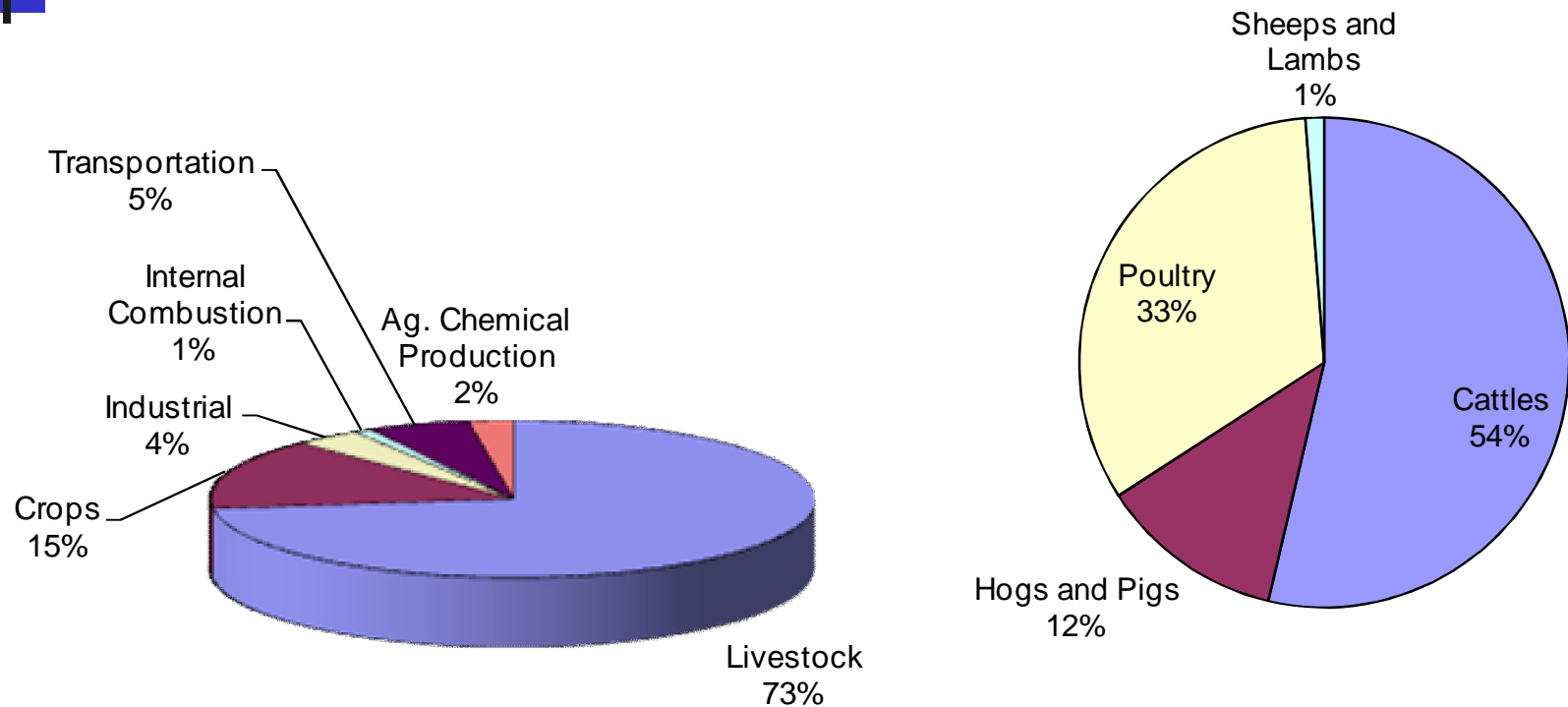
## ■ Manure Handling

- ★ Manure storage time
- ★ Diffusive and convective transport in the manure



# **Ammonia Emissions from Poultry Facilities**

# How Much NH<sub>3</sub> is Emitted?



<sup>1</sup> <http://pubwiki.extension.org/mediawiki/files/7/70/08junPPmartin.pdf>

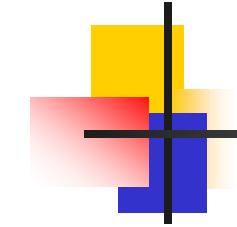
<sup>2</sup> USEPA, 2002. [http://www.epa.gov/ttn/chief/ap42/ch09/related/nh3inventorydraft\\_jan2004.pdf](http://www.epa.gov/ttn/chief/ap42/ch09/related/nh3inventorydraft_jan2004.pdf)

# High-rise deep-pit (HR) layer barn

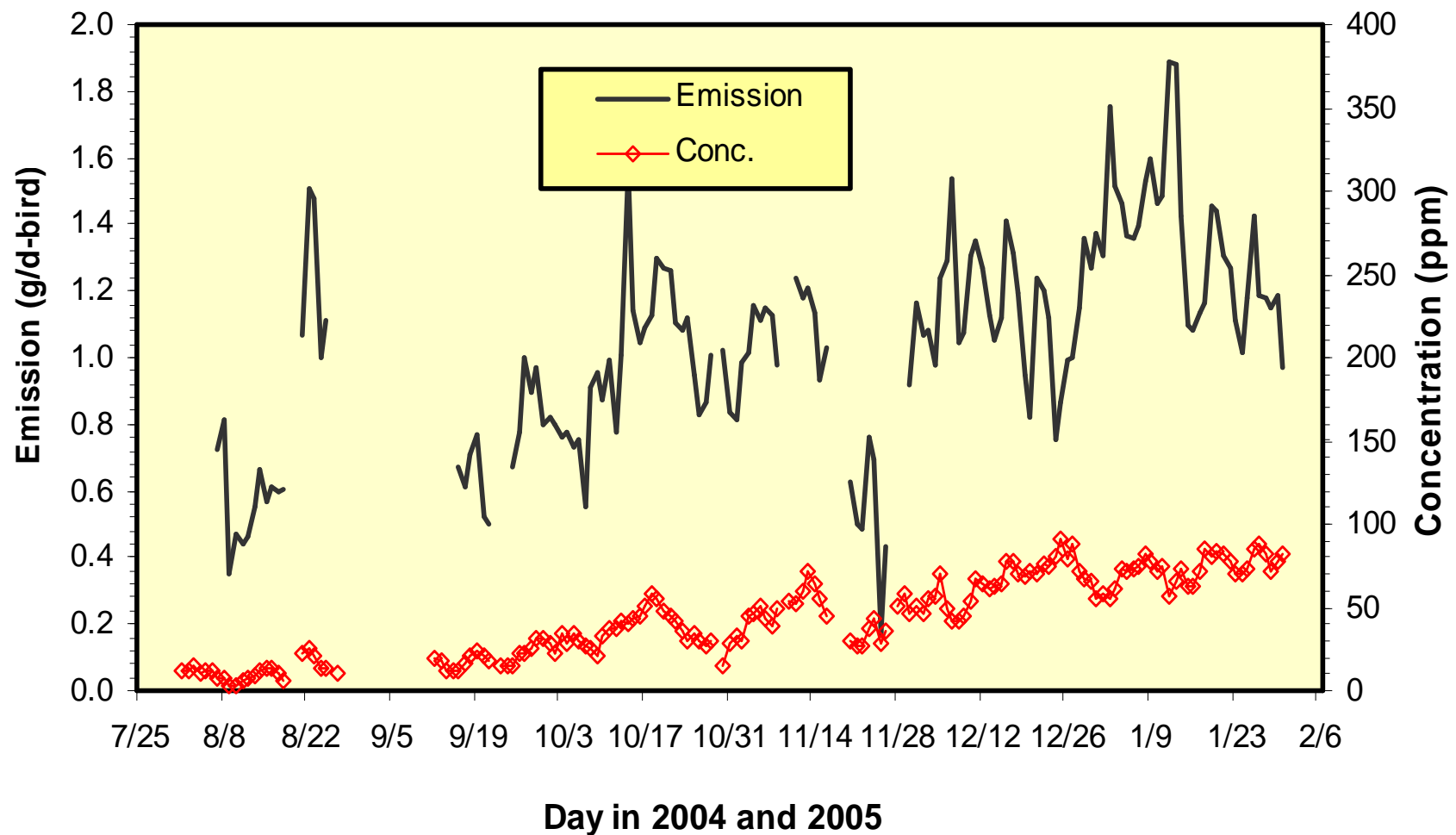




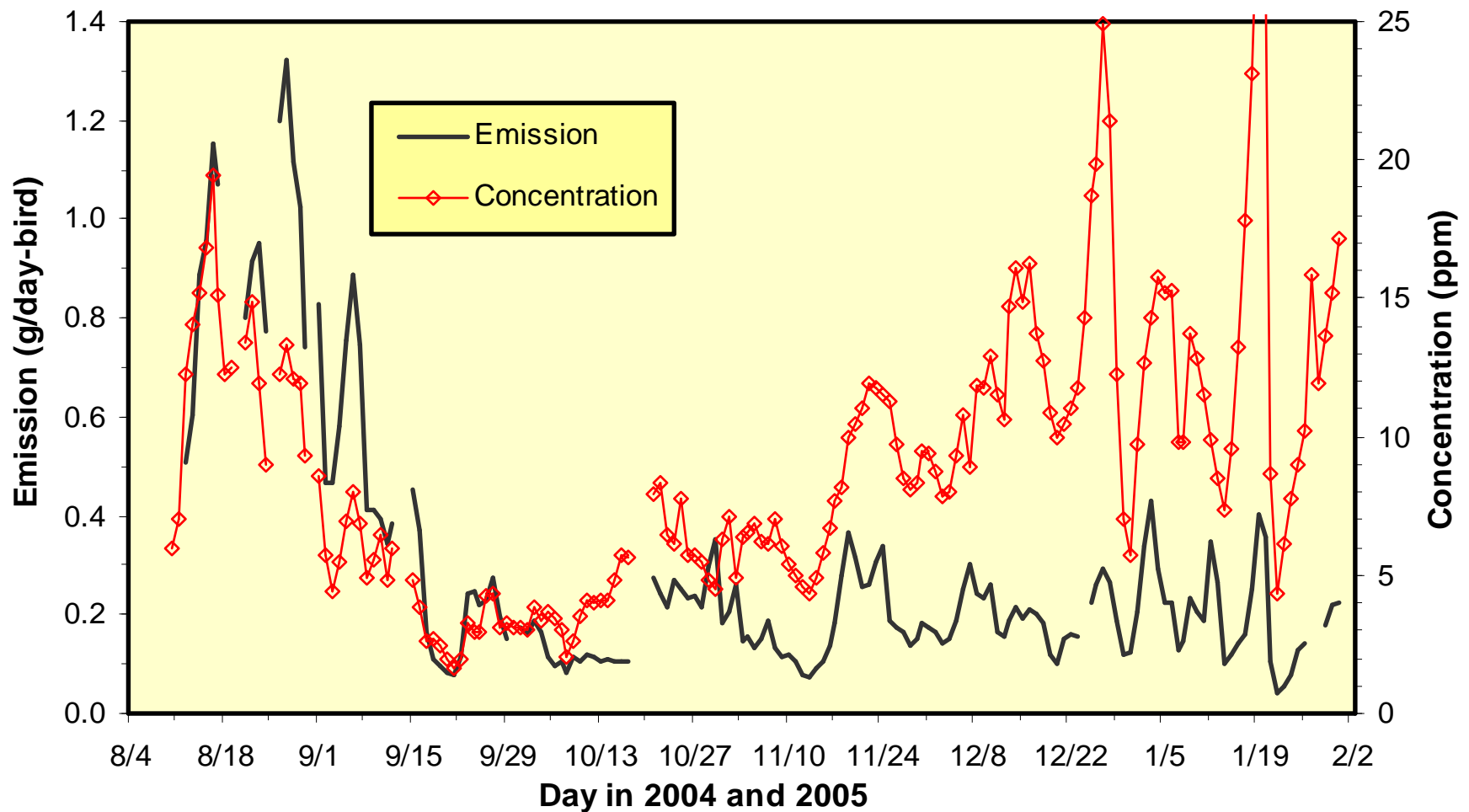
# Manure-belt (MB) layer barns



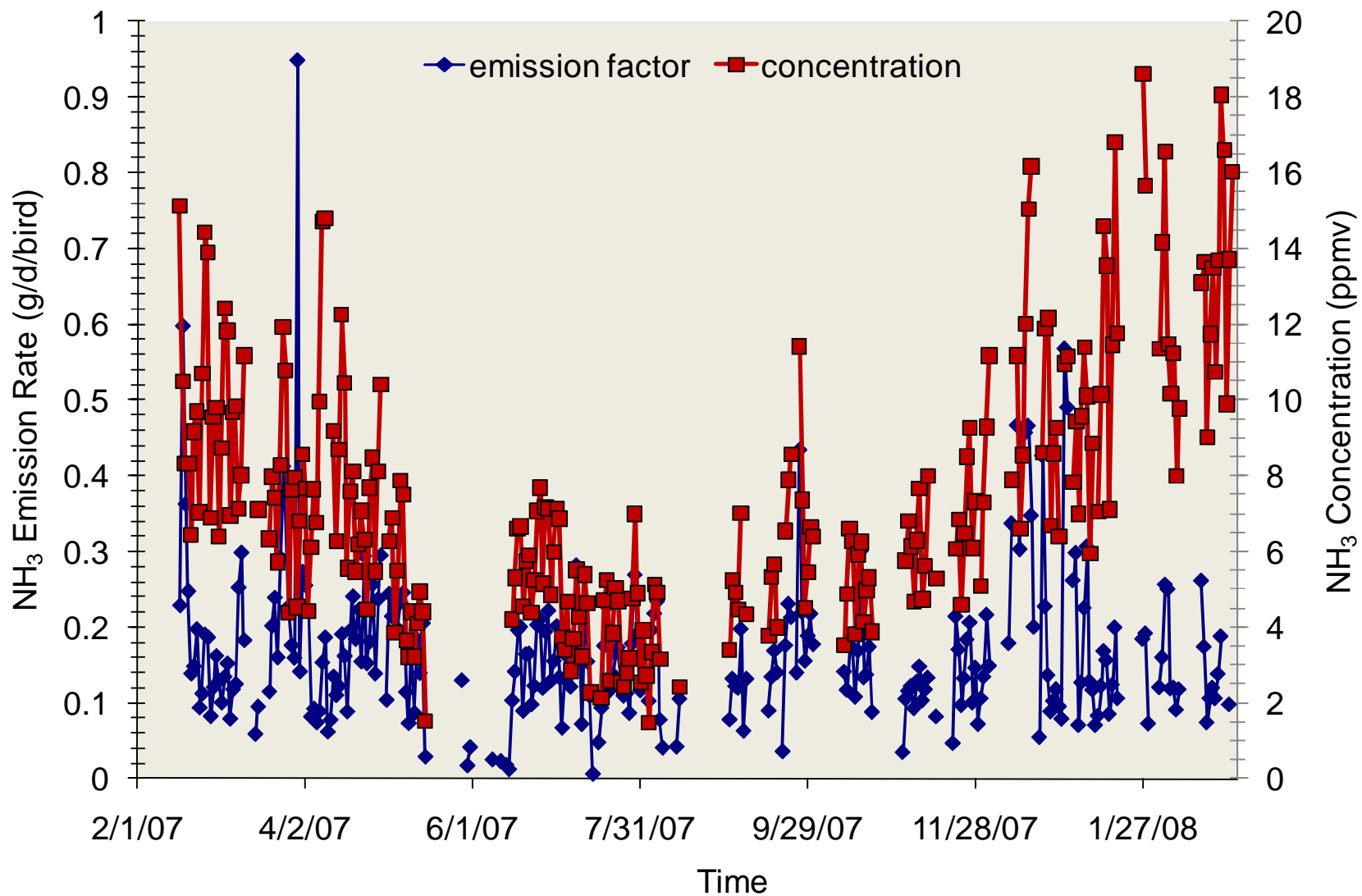
# Deep-Pit High-Rise Barn- NH<sub>3</sub> Concentration & Emission



# Manure-Belt Barn— NH<sub>3</sub> Concentration & Emission

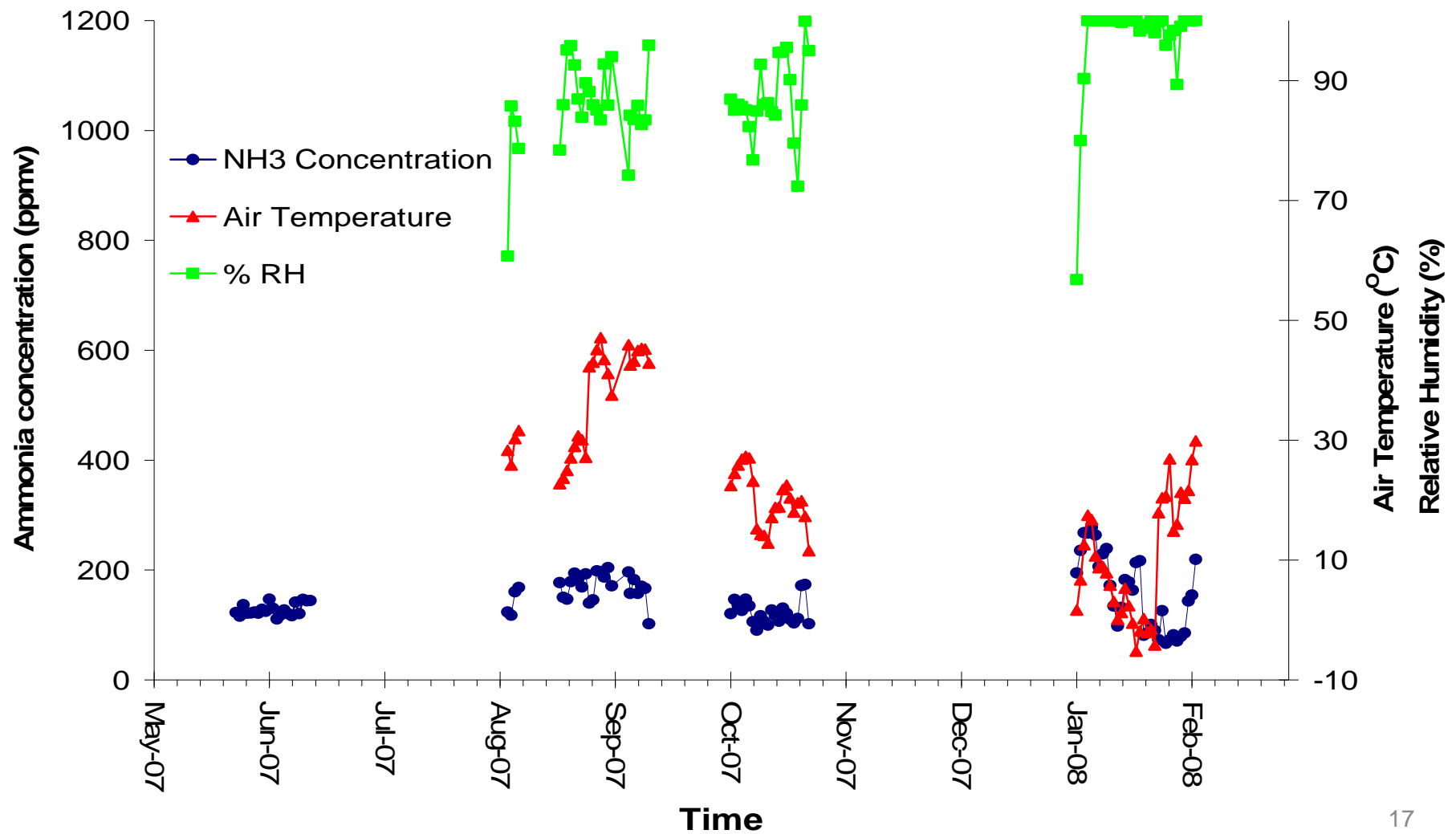



# Ammonia Concentrations & Emissions from Manure-Belt Layer Facilities





# Seasonal Variations in Ammonia Concentrations and Emissions





# A Summary of NH<sub>3</sub> Concentrations and Emissions of Layer Facilities

Poultry Layer Facilities	Daily Average NH <sub>3</sub> Concentration (ppmv) Mean (Min-Max)	Daily Average NH <sub>3</sub> Emissions (g/d/bird) Mean ± Std
High-Rise Deep-Pit Barns	40 (0-90)	1.03 ± 0.39
Manure-Belt Barns 1— daily 1/7 manure removal	9.1 (0-25)	0.29 ± 0.27
Manure-Belt Barn 2— manure removal every 3 day	6.7 (0-19)	0.16 ± 0.13
Manure Composting Facilities	130 (66-278)	0.32 ± 0.14

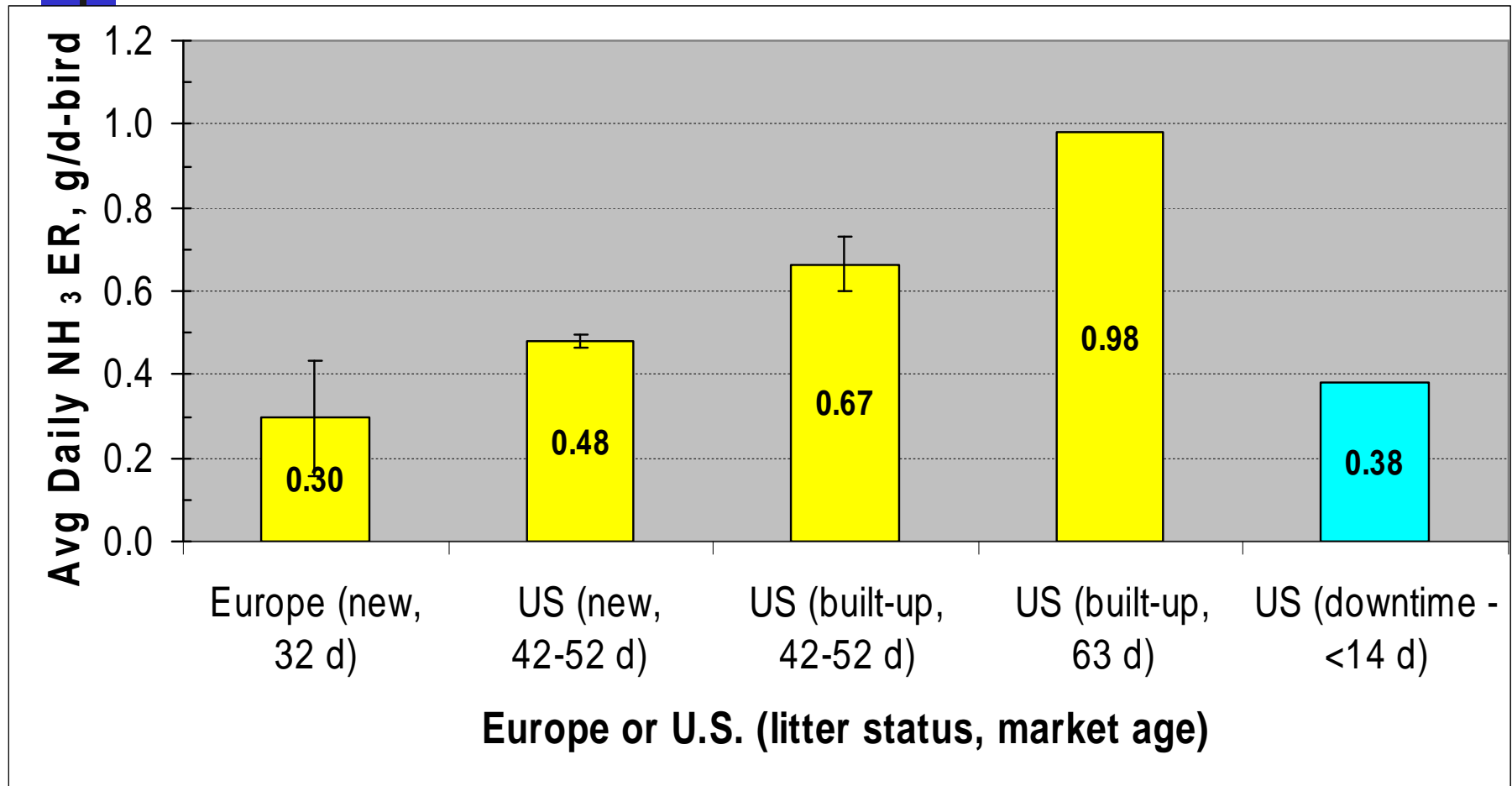
# U.S. Broiler & Turkey Facilities

- Mostly built-up litter, some with litter treatment
- Tunnel ventilation or hybrid system
- Cooling pads or misters
- Pancake brooders & space heaters
- Broilers marketed at 6-9 wk
- Hen & tom turkeys at 12 & 20 wk



Source: Hongwei Xin, 2011

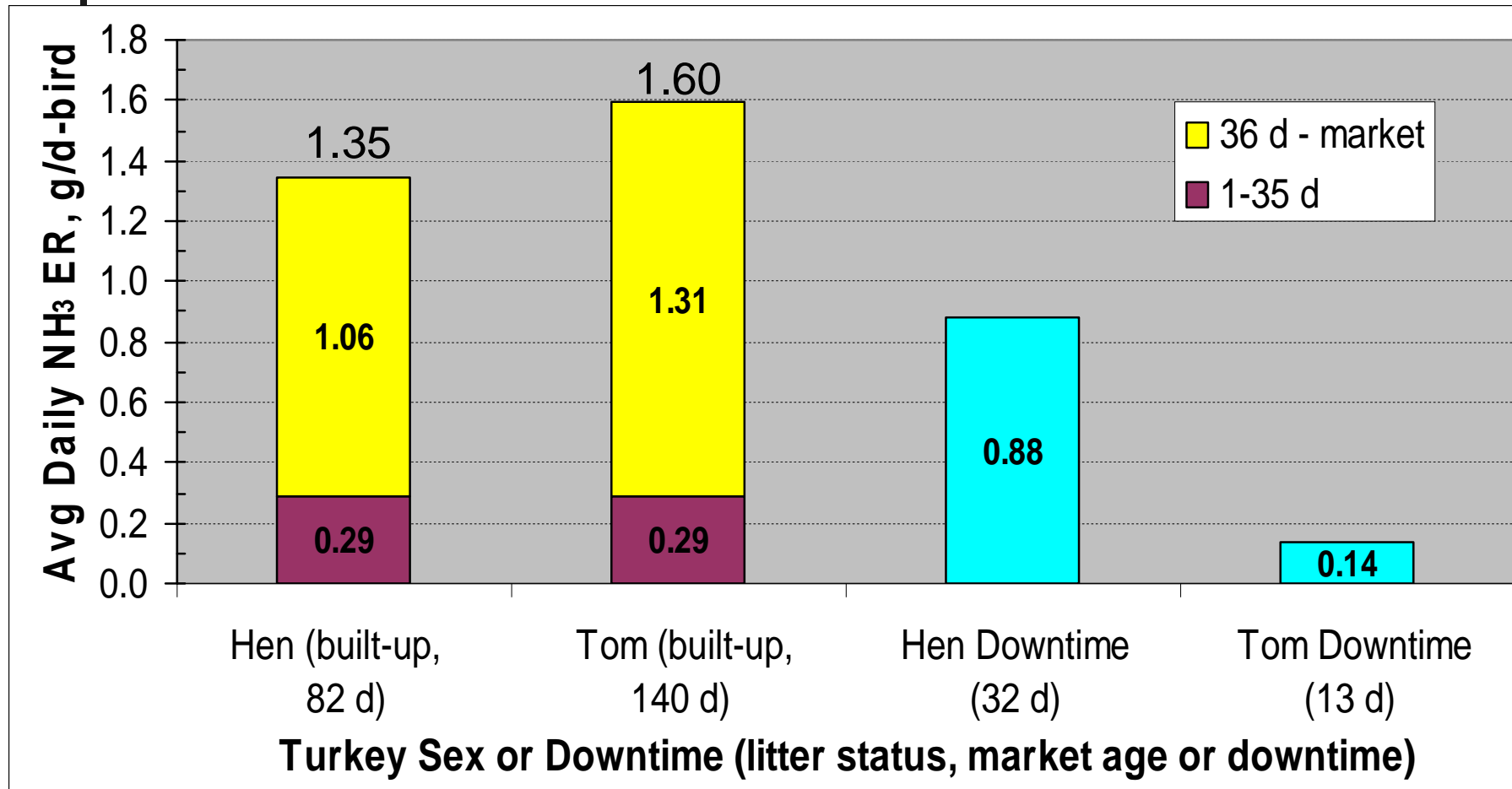
# NH<sub>3</sub> Emission Rate (ER) of Broiler Houses – European & US Data



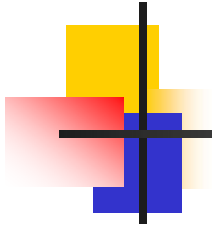
Source: Hongwei Xin, 2011

# NH<sub>3</sub> Emission Rate (ER) of Turkey Houses in Iowa & Minnesota

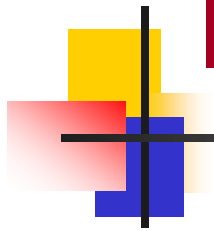
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Source: Hongwei Xin, 2011



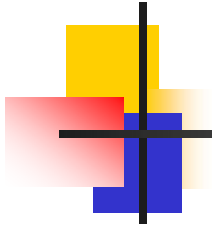
# **Health and environmental impacts of Ammonia Emission**



# Its impact on Human health

The health effects of ammonia have been recognized through extensive literature.

Concentration (ppm)	Length of exposure	Health effects
50	< 24 hours	Slight, temporary eye and throat irritations and urge to cough
100	6 weeks	Irritation of eyes, nose, and throat
500	30 mins	Increased air intake into lungs, sore nose and throat
5000	< 30 mins	Kills quickly



# Impacts on Poultry Animal Health

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- ★ At level of 10 ppm, trachea irritation was shown in turkeys.
- ★ At above 25 ppm, growth rate and feed conversion was impaired and the final body weight was reduced.
- ★ At levels of above 50 ppm, the birds are more susceptible to bacterial infections such as *E. coli*.
- ★ At above 100 ppm, the chick mortality was increased significantly.





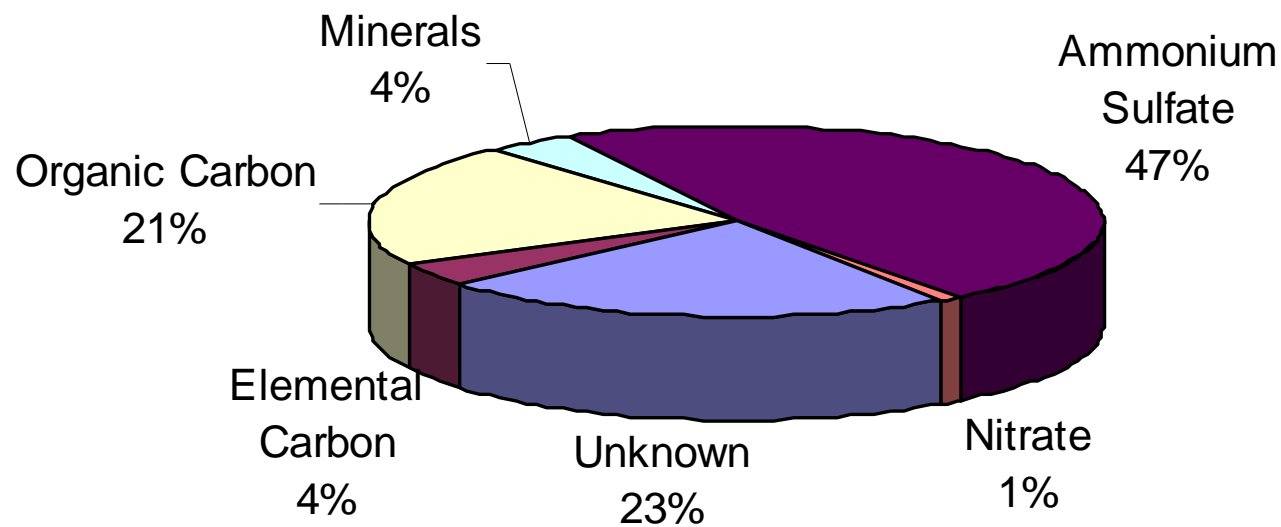
# NH<sub>3</sub> and Particulate Matter<sup>6</sup>

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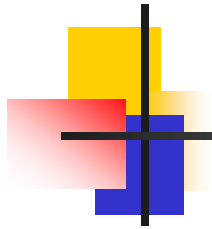
Ammonia gas can react with sulfuric and nitric acids in the atmosphere to form fine particles

- ◆  $\text{NH}_3(g) + \text{H}_2\text{SO}_{4(g)} \rightarrow (\text{NH}_4)_2\text{SO}_{4(g,l)}$  (ammonium sulfate)
- ◆  $\text{NH}_3(g) + \text{HNO}_{3(g)} \rightarrow \text{NH}_4\text{NO}_{3(g,l)}$  (ammonium nitrate)

# NH<sub>3</sub> and PM<sub>2.5</sub>



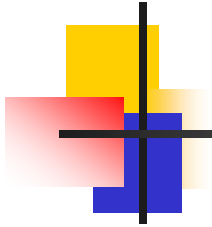
Source: Anderson N., R. Strader, and C. Davidson. 2003. Airborne reduced nitrogen: ammonia emissions from agriculture and other sources. *Environment International*. 29(2003): 277-286.



## Its Environmental Impacts Cont...

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- ◆ Ammonia deposition or wet deposition to the ecosystem leads to over-fertilization or eutrophication.
- ◆ The nutrient imbalance in the soil can be harmful to some crops while excessive nutrients in aquatic systems lead to decreased biological diversity;
- ◆ Ammonium sulfates deposited into soil and water systems can also get oxidized to form a mixture of nitric and sulfuric acids to cause ecosystem acidification.
- ★ Conversion to ammonium ion, and subsequently  $\text{PM}_{2.5}$ , its aerosol form, and eventual smog formation.



# **Overview of Effective BMPs and BATs for Ammonia Emissions from Poultry Facilities**

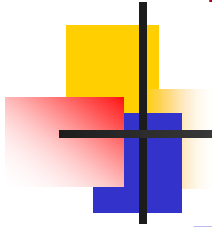


# Principles for Mitigating NH<sub>3</sub> Emissions in Animal Buildings

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- Minimize air emission generation at the sources
- Reduce ammonia emission in the building
- Reduce ammonia emission at manure storages
- Capture ammonia emission at the exhausts of poultry facilities

# Factors Affecting $\text{NH}_3$ Emission and Transport



## ■ Dietary Factors

- ★ Excess crude protein in diet
- ★ Low conversion of dietary N to animal products
- ★ Total nitrogen content

## ■ Environmental Factors

- ★ pH
- ★ Temperature
- ★ Moisture content
- ★ Chemical and microbiological activities--additives
- ★ Ammonia concentration of air
- ★ Ventilation air / wind speed

## ■ Manure Handling

- ★ Manure storage time
- ★ Diffusive and convective transport in the manure



# Diet Manipulation and Feed Additives

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One of the most important means of mitigation of air emissions

- Reducing nitrogen intake and using synthetic amino acid to reduce  $\text{NH}_3$  emission.
- Adding dietary fiber to reduce ammonia emission (Xin, 2011)
- Balancing feed ingredients and improving the digestibility to reduce odor and ammonia emissions.



# Manure Additives for Odor and Gas Control

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Chemical and biological substances added to manure that

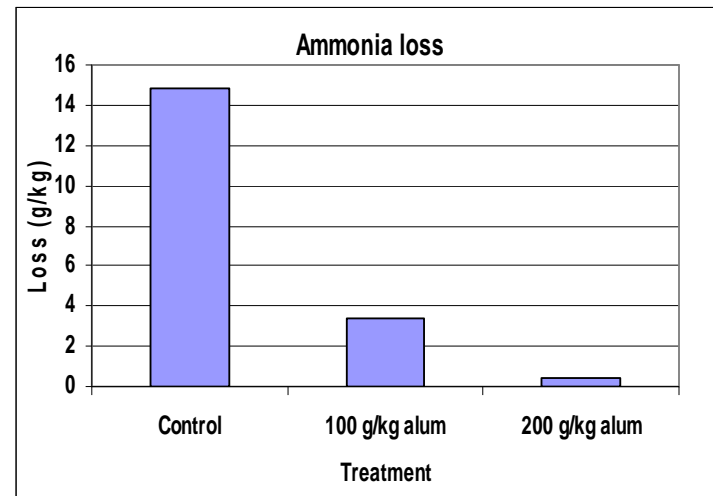
- Change manure pH: Alum (liquid or powder), ferric sulfate, sodium bisulfate
- Adsorb ammonium: Zeolite
- Change manure microbial population: many products on market

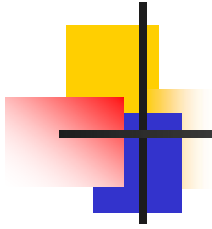


# Alum has been applied to Layer Manure and Broiler Litter

Alum has been shown to reduce ammonia volatilization very significantly during a 42-day incubation of poultry litter

- A field test at a Ohio layer facility showed that Aluminum Sulfate (Liquid Alum) and Aluminum Chloride ( $\text{AlCl}_3$ ) spraying system reduced  $\text{NH}_3$  emission by 33%- 40% in short test periods.





# Good Manure Handling Practice

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- Keep Manure dry
- Remove manure out of the building
- Store manure in cool, covered facilities
- Reduce manure storage surface area

# Poultry Facilities and $\text{NH}_3$ Emissions



- High-Rise Deep-Pit Layer Facility





# Poultry Facilities and $\text{NH}_3$ Emissions



**Manure-Belt (MB)  
layer facility**



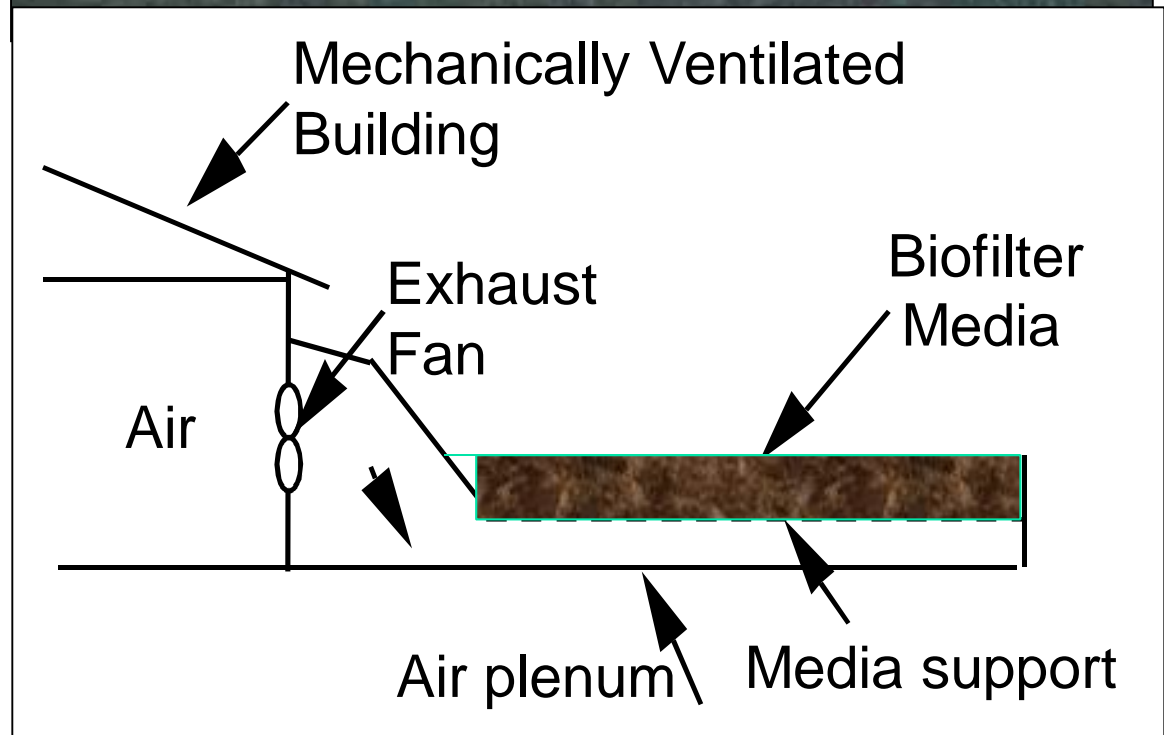
# Manure Composting

- Two times nitrogen conservation rate than the deep-pit high-rise layer facilities (Keener et al. 2002)
- High ammonia concentrations in the composting building



# Biofilter

- Odor reduction  
80% - 90%
- $\text{H}_2\text{S}$  reduction  
80% - 90%
- $\text{NH}_3$  reduction  
50% - 60%

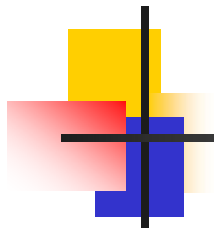


# Washing Walls

Use water to “scrub” air leaving buildings. Water recirculates through evaporative pad scrubber. Used with power ventilation systems

- Total dust reduced
  - ★ 20% - 60% reduction
- Ammonia reduced
  - ★ 33% - 50% reduction
- Odors were slightly reduced





# Wet Scrubbers

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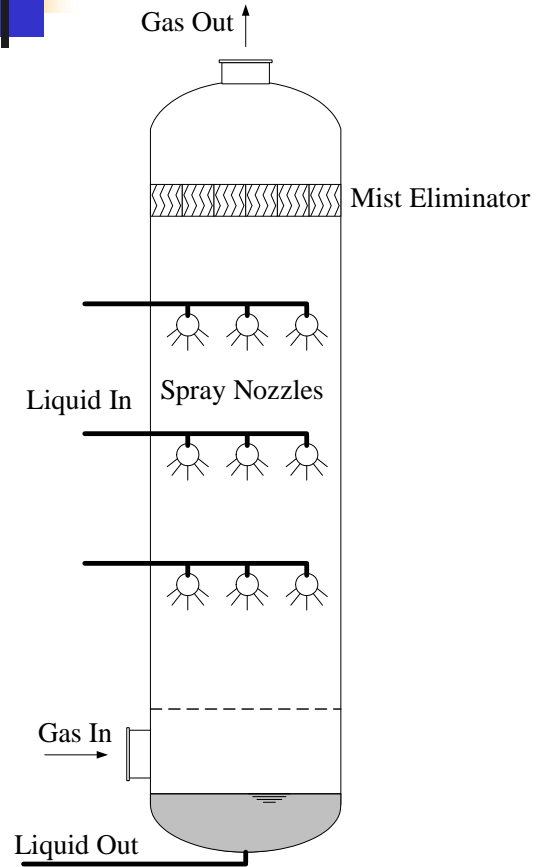
Use water and chemicals to “scrub” air pollutants laden air streams leaving buildings

Two types of Wet Scrubbers:

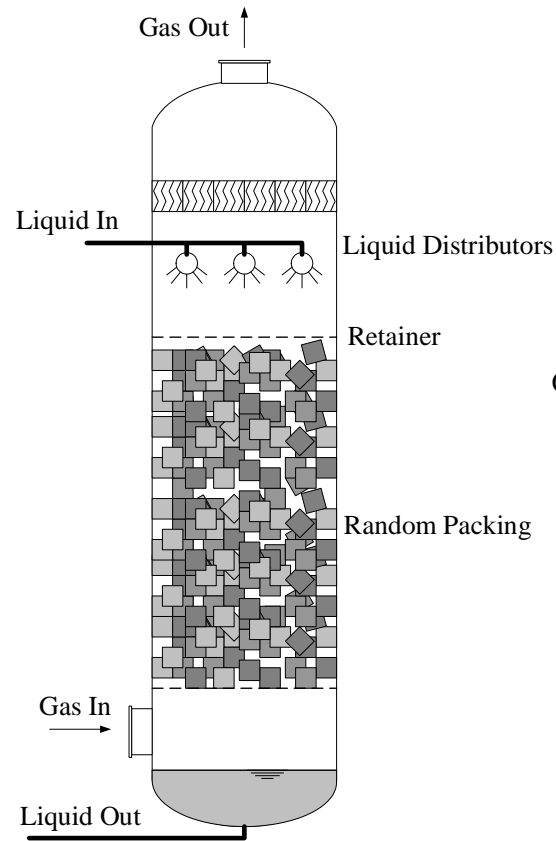
- European Acid packed bed wet scrubber or fiber bed acid scrubber— ammonia removal efficiencies of 63%-98% with an average of 81% (Melse et al., 2005, 2008)
- OSU acid spray wet scrubbers can achieve  $\text{NH}_3$  removal efficiencies of 99%-75% at 5-400 ppm inlet  $\text{NH}_3$  concentration (Manuzon et al. 2007)



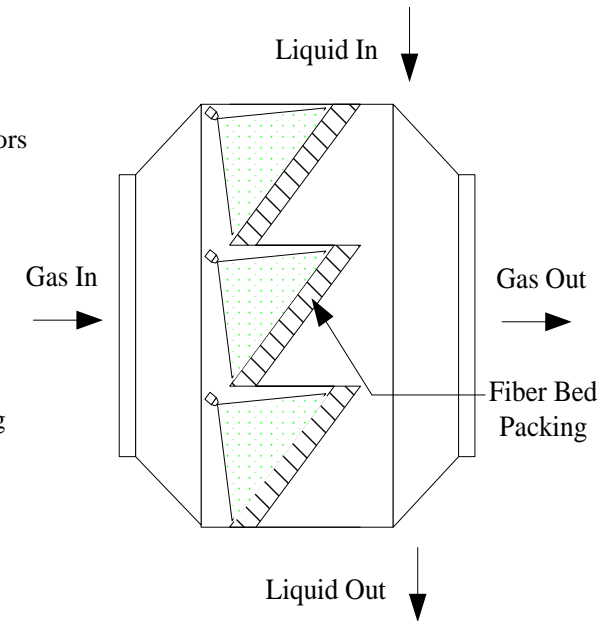
# Schematics of Wet Scrubbers



Spray

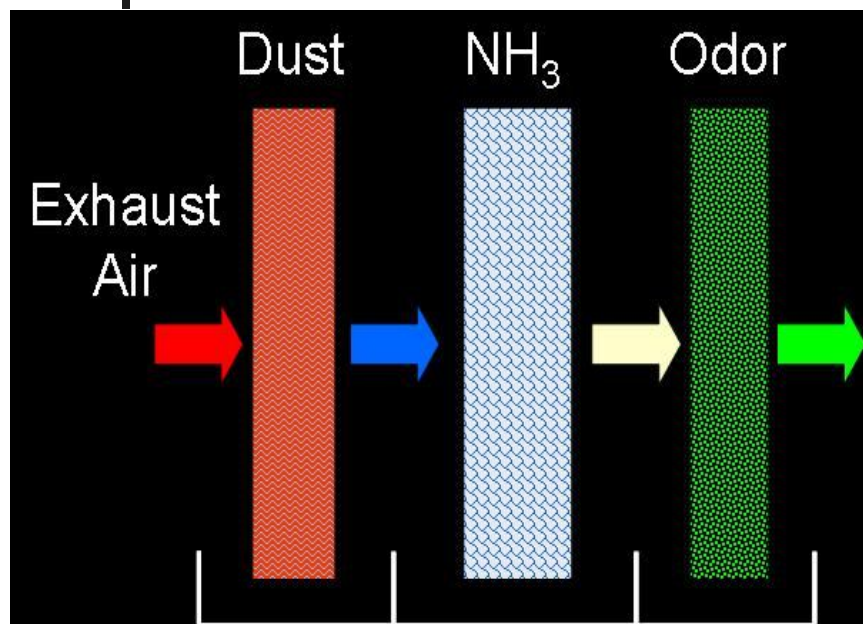


Packed-Bed



Fiber-Bed

# European Fiber Bed Wet Scrubbers



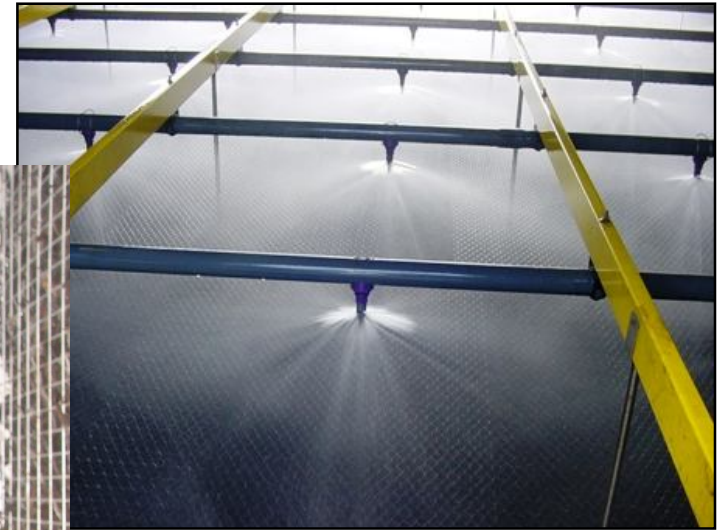
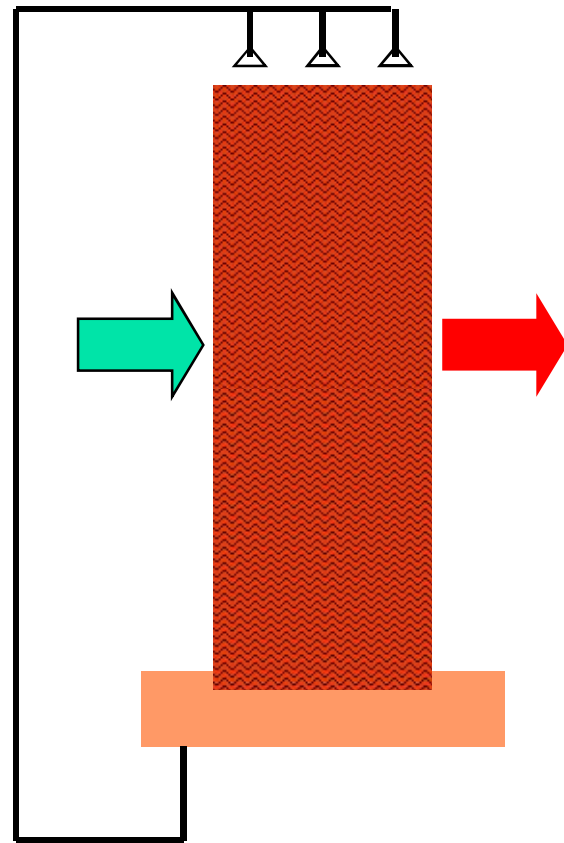
Packed-bed and 3-stage scrubber :

- Capital cost: \$47-\$72/pig
- Operation cost:\$15-\$19/year



Source: Hongwei Xin, 2011

# Ammonia Removal

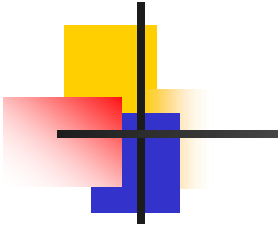


- Sulfuric acid solution pH 2
- Recirculation
- Efficiency 63-98%

Source: Hongwei Xin, 2011



# OSU Spray Acid Wet Scrubber



## **Spray scrubbers**

- spray system has low back pressure
- scrubbing liquid recycled as fertilizer



# Summary Points...

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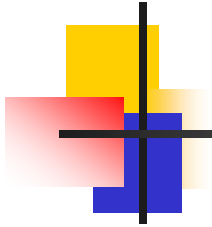
- ◆ Inefficient conversion of feed N to animal products resulted in N excretion as ammonia.
- ◆ Total nitrogen in manure, temperature, pH, manure moisture level, and microbial activities affect ammonia generation. Air speed, ammonia concentration, and weather conditions affect ammonia dispersion and transport
- ◆ High level of ammonia has significant health effects on human and poultry animals
- ◆ Ammonia emission is a pre-cursor of PM<sub>2.5</sub> particle and cause ecosystem eutrophication and acidity.



## Summary Points (Cont.)

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- Effective BMPs and BATs to reduce NH<sub>3</sub> emissions from poultry facilities are:
  - Diet manipulation and feed additives
  - Manure additives
  - Drying manure
  - Storing manure in cool and covered space with small manure surface area
  - Composting manure
  - Adopting manure belt building systems
  - Using wet scrubbers to capture ammonia for fertilizer



# Thank You !      Questions?

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